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In chemistry, a solution is a special type of homogeneous mixture composed of two or more substances. In such a mixture, a solute is a substance dissolved in another substance, known as a solvent. The mixing process of a solution happens at a scale where the effects of chemical polarity are involved, resulting in interactions that are specific to solvation.

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## **Solution (chemistry) - Wikipedia**

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The bottom organic layer was then drained from the funnel and set aside. This was repeated two more times with two 10 mL

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portions of dichloromethane. Once again, caution was taken to not allow the pressure to build up within the separatory funnel. The organic layer was then transferred into an Erlenmeyer flask and dried with anhydrous sodium ...

### **Organic Chemistry Lab Report—Synthesis of Cyclohexanone ...**

Tests for the Functional Groups Present in Organic Compounds  
Molecules of organic compounds except that of hydrocarbons can be divided into two parts, a reactive part which is referred to as functional group and a skeleton of carbon atoms called alkyl group. The properties of a compound are largely determined by the functional group. Different [...]

### **Tests for the Functional Groups Present in Organic ...**

Solvent, substance, ordinarily a liquid, in which other materials dissolve to form a solution. Polar solvents (e.g., water) favor

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formation of ions; nonpolar ones (e.g., hydrocarbons) do not. Solvents may be predominantly acidic, predominantly basic, amphoteric (both), or aprotic (neither).

## **solvent | Definition, Examples, & Facts | Britannica**

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Sulfur (in nontechnical British English: sulphur) is a chemical element with the symbol S and atomic number 16. It is abundant, multivalent and nonmetallic. Under normal conditions, sulfur atoms form cyclic octatomic molecules with a chemical formula S<sub>8</sub>. Elemental sulfur is a bright yellow, crystalline solid

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at room temperature. Sulfur is the tenth most common element by mass in the universe ...

## **Sulfur - Wikipedia**

The stripped organic phase from the stripping section and the alkaline solution with the controlled flow ratio and flow rate directly flow into the mixer of the saponification cell, where the saponification reaction takes place, and the mixed phase flows into settler to separate phase. The organic phase in the settler is the saponified organic ...

## **Saponification - an overview | ScienceDirect Topics**

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