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Aqueous Solutions dissociation is the separation of ions that occurs when an ionic compound dissolves see sample problem A on page 436 do practice problem #1 on page 436 Solubility precipitation reactions occur when combinations of ions in a solution have an extremely low solubility and a precipitate forms see table 1 on page

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437 of the textbook for general solubility guidelines Net ...

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Vocabulary. dissociation. net ionic equation. spectator ions. ionization. the separating of a molecule into simpler



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molecules, atoms, rad.... an equation that includes only those compounds and ions that u.... ions that are present in a solution in which a reaction is tak....  
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Chapter 13: ... Answers Chapter 12

SECTION 1 SHORT ANSWER 1 c 2 a 3 b 2

a alcohol b water c the gels 3 The mixture is a colloid The properties are consistent with those reported in

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priscillalogan. Chemistry Chapter 13  
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spectator ions. ionization. the separation

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of ions that occurs when an ionic compound dissolves. This includes only those compounds and ions that undergo a chemical change. Ions that do not take part in a chemical reaction and are found in the solution are called spectator ions.

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Chemistry. 0% average ... 13 Qs .  
Animals . 33.1k ...

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Modern Chemistry 6 The Periodic Law  
CHAPTER 5 REVIEW The Periodic Law  
SECTION 3 SHORT ANSWER Answer the

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following questions in the space provided. 1. \_\_\_\_\_ When an electron is added to a neutral atom, energy is (a) always absorbed.

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#### **in Chemistry - Chemistry**

CHAPTER 6 REVIEW Chemical Bonding

SECTION 3 SHORT ANSWER Answer the

following questions in the space

provided. 1. a The notation for sodium

chloride, NaCl, stands for one (a)

formula unit. (c) crystal. (b) molecule.

(d) atom. 2. d In a crystal of an ionic

compound, each cation is surrounded by

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a number of (a) molecules. (c) dipoles.  
(b ...

## **6 Chemical Bonding**

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SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. Determine whether each of the following is an example of observation and data, a theory, a hypothesis, a control, or a model.

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chapter 10 section Chapter 10 Chemical Calculations and equations 10, that can be made from  $109 \times 10^4$  kilograms of phosphorus in the second of these three reactions The answer is  $250 \times 10^4$  kg P  
40 10 We used the following steps:

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