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charge of monoatomic (1 atom) ion
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Study the electron dot diagrams for lithium, carbon, flourine, and neon in figure 6-1. Choose the statement that correctly identifies the most stable of

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the elements. A) lithium is the most stable element because it has to lose only one electron to achieve a stable configuration. B) carbon is the most stable element because it can form four bonds.

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Section 6.4 The Structure of Metals (pages 176–181) This section discusses metallic bonds and the properties of metals. It also explains how the properties of an alloy are controlled. <http://mrmicco.weebly.com/uploads/1/2/5/8/12584959/psgs1064b.pdf>

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